

# **MSDS OF CALCIUM CHLORIDE**

#### 1. Product Identification

Synonyms: Calcium Chloride Anhydrous

CAS No.: 10043-52-4 Molecular Weight: 147.02 Chemical Formula: CaCl2

Emergency Contact: CHEMTRAC +44(0)1524 510278

#### 2. Composition/Information on Ingredients

Ingredient	CAS No	Percent	Hazardous
Calcium Chloride	10043-52-4	94 - 100%	Yes

#### 3. Hazards Identification

# **Emergency Overview**



WARNING! CAUSES IRRITATION TO SKIN, EYES AND RESPIRATORY TRACT. HARMFUL IF SWALLOWED OR INHALED.

SAF-T-DATA<sup>(tm)</sup> Ratings (Provided here for your convenience)

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Health Rating: 1 - Slight

Flammability Rating: 0 - None Reactivity Rating: 0 - None Contact Rating: 2 - Moderate

Lab Protective Equip: GOGGLES; LAB COAT Storage Color Code: Orange (General Storage)

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# **Potential Health Effects**

\_\_\_\_\_

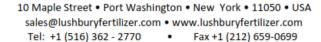
## Inhalation:

Granular material does not pose a significant inhalation hazard, but inhalation of dust may cause irritation to the respiratory tract, with symptoms of coughing and shortness of breath.

## Ingestion:

Low toxicity material but ingestion may cause serious irritation of the mucous membrane due to heat of hydrolysis. Large amounts can cause gastrointestinal upset, vomiting, abdominal pain.

#### **Skin Contact:**





Solid may cause mild irritation on dry skin; strong solutions or solid in contact with moist skin may cause severe irritation, even burns.

**Eye Contact:** H319 Causes serious eye irritation.

Hazard may be either mechanical abrasion or, more serious, burns from heat of hydrolysis and chloride irritation.

# **Chronic Exposure:**

No information found.

# **Aggravation of Pre-existing Conditions:**

No information found.

#### 4. First Aid Measures

#### Inhalation:

Remove to fresh air. If not breathing, give artificial respiration. If breathing is difficult, give oxygen. Get medical attention.

#### Ingestion:

Induce vomiting immediately as directed by medical personnel. Never give anything by mouth to an unconscious person. Get medical attention.

#### **Skin Contact:**

Wipe off excess material from skin then immediately flush skin with plenty of water for at least 15 minutes. Remove contaminated clothing and shoes. Get medical attention. Wash clothing before reuse. Thoroughly clean shoes before reuse.

# **Eye Contact:**

Immediately flush eyes with plenty of water for at least 15 minutes, lifting lower and upper eyelids occasionally. Get medical attention immediately.

#### **Note to Physician:**

Oral ingestion may cause serum acidosis.

# 5. Fire Fighting Measures

#### Fire:

Not considered to be a fire hazard.

#### **Explosion:**

Not considered to be an explosion hazard.

## Fire Extinguishing Media:

Use any means suitable for extinguishing surrounding fire.

## **Special Information:**

In the event of a fire, wear full protective clothing and NIOSH-approved self-contained breathing apparatus with



full face piece operated in the pressure demand or other positive pressure mode.

#### 6. Accidental Release Measures

Ventilate area of leak or spill. Wear appropriate personal protective equipment as specified in Section 8. Spills: Sweep up and containerize for reclamation or disposal. Vacuuming or wet sweeping may be used to avoid dust dispersal. Small amounts of residue may be flushed to sewer with plenty of water.

#### 7. Handling and Storage

Keep in a tightly closed container, stored in a cool, dry, ventilated area. Protect against physical damage. Moist calcium chloride and concentrated solutions can corrode steel. When exposed to the atmosphere, calcium chloride will absorb water and form a solution. Containers of this material may be hazardous when empty since they retain product residues (dust, solids); observe all warnings and precautions listed for the product.

#### 8. Exposure Controls/Personal Protection

#### **Airborne Exposure Limits:**

None established.

#### **Ventilation System:**

A system of local and/or general exhaust is recommended to keep employee exposures as low as possible. Local exhaust ventilation is generally preferred because it can control the emissions of the contaminant at its source, preventing dispersion of it into the general work area. Please refer to the ACGIH document, *Industrial Ventilation, A Manual of Recommended Practices*, most recent edition, for details.

# Personal Respirators (NIOSH Approved):

For conditions of use where exposure to the dust or mist is apparent, a half-face dust/mist respirator may be worn. For emergencies or instances where the exposure levels are not known, use a full-face positive-pressure, air-supplied respirator. WARNING: Air-purifying respirators do not protect workers in oxygen-deficient atmospheres.

**Skin Protection:** P280 Wear protective gloves/protective clothing/eye protection/face protection. Wear protective gloves and clean body-covering clothing.

#### **Eve Protection:**

P305+P351+P338 IF IN EYES: Rinse cautiously with water for several minutes. Remove contact lenses, if present and easy to do. Continue rinsing.

Use chemical safety goggles and/or full face shield where dusting or splashing of solutions is possible. Maintain eye wash fountain and quick-drench facilities in work area.

#### **Other Control Measures:**

Maintain good housekeeping in work area. Dust deposits on floors and other surfaces may pick up moisture and cause the surfaces to become slippery and present safety hazards.

# 9. Physical and Chemical Properties

**Appearance** 



Coarse white Flakes or mixture of coarse white powder and medium sized granules or flakes.

Odor: Odorless.

**Solubility:** Freely soluble in water with heat liberation.

**Density:** 0.835

Ph: 8 - 9 Aqueous solution

% Volatiles by volume @ 21C (70F): 0

**Boiling Point:** > 1600C (> 2912F)

Melting Point: ca. 260C (ca. 500F)

Vapor Density (Air=1): No information found.

Vapor Pressure (mm Hg): No information found.

Evaporation Rate (BuAc=1): No information found.

# 10. Stability and Reactivity

# Stability:

Stable under ordinary conditions of use and storage. Substance will pick up moisture from the air and go into solution if exposed in open containers.

# **Hazardous Decomposition Products:**

Emits toxic chlorine fumes when heated to decomposition. May form hydrogen chloride in presence of sulfuric or phosphoric acids or with water at elevated temperatures.

#### **Hazardous Polymerization:**

Will not occur.

# Incompatibilities:

Methyl vinyl ether, water, zinc, bromine trifluoride, mixtures of lime and boric acid, barium chloride, and 2-furan percarboxylic acid. Metals will slowly corrode in aqueous calcium chloride solutions. Aluminum (and alloys) and yellow brass will be attacked by calcium chloride.

#### **Conditions to Avoid:**

Incompatibles.

#### 11. Toxicological Information

For calcium chloride anhydrous: oral rat LD50: 1000 mg/kg; investigated as a mutagen, tumorigen.



10 Maple Street • Port Washington • New York • 11050 • USA sales@lushburyfertilizer.com • www.lushburyfertilizer.com
Tel: +1 (516) 362 - 2770 • Fax +1 (212) 659-0699

\Cancer Lists\					
		NTP Carcinogen			
Ingredient	Known	Anticipa	ited IA	RC Category	
Calcium Chloride (10043-52-4)	No	No	1	 None	
12. Ecological Information					
<b>Environmental Fate:</b> Based on available information for Calc	cium Chloride	e anhydrous	s, this materi	al will not biodegrade or bioaccumu	late
<b>Environmental Toxicity:</b> Based on data for Calcium Chloride and	hydrous, the	LC50/96-ho	our values for	r fish are over 100 mg/l.	
13. Disposal Considerations					
Whatever cannot be saved for recover disposal facility. Processing, use or constate and local disposal regulations may contents in accordance with federal, st	ontamination ay differ from	of this pron of federal dis	oduct may cl sposal regula	nange the waste management option	ons.
14. Transport Information					
Not regulated.					
15. Regulatory Information					
\Chemical Inventory Status	- Part 1\				
Ingredient	TSCA	. EC	Japan	Australia	
Calcium Chloride (10043-52-4)	Yes	Yes	Yes	Yes	
\Chemical Inventory Status	- Part 2\				

--Canad--



Ingredient 	Korea	DSL	NDSL	Phil.						
Calcium Chloride (10043-52-4)		Yes	No \	⁄es						
\Federal, State & International Regulations - Part 1\										
	-SARA 3	313								
Ingredient	RQ TP	Q List	C	hemical Ca	itg.					
Calcium Chloride (10043-52-4)	 No No	No	N	0						
\Federal, State & Internatio	onal Regulati	ons - Part 2	·\							
		-RCRA	TS	CA-						
Ingredient	CERCLA	261.33	8	(d)						
Calcium Chloride (10043-52-4)	 No	No	1	 No						
Chemical Weapons Convention:	No TS	CA 12(b):	No	CDTA: N	No					
SARA 311/312: Acute: Yes	Chronic: N	o Fire: N	No Pres	sure: No						
Reactivity: No (Mixtui	re / Solid)									

# WHMIS:

Australian Hazchem Code: No information found.

Poison Schedule: No information found.

This MSDS has been prepared according to the hazard criteria of the Controlled Products Regulations (CPR) and the MSDS contains all of the information required by the CPR.



#### 16. Other Information

NFPA Ratings: Health: 1 Flammability: 0 Reactivity: 1

#### **Label Hazard Warning:**

WARNING! CAUSES IRRITATION TO SKIN, EYES AND RESPIRATORY TRACT. HARMFUL IF SWALLOWED OR INHALED.

#### **Label Precautions:**

Avoid contact with eyes, skin and clothing. Wash thoroughly after handling. Avoid breathing dust. Keep container closed. Use only with adequate ventilation.

#### **Label First Aid:**

If swallowed, induce vomiting immediately as directed by medical personnel. Never give anything by mouth to an unconscious person. In case of contact, wipe off excess material from skin then immediately flush eyes or skin with plenty of water for at least 15 minutes. Remove contaminated clothing and shoes. Wash clothing before reuse. If inhaled, remove to fresh air. If not breathing, give artificial respiration. If breathing is difficult, give oxygen. In all cases, get medical attention.

#### **Product Use:**

Laboratory Reagent.

#### **Revision Information:**

Mixture. New 16 section MSDS format, all sections have been revised.

This MSDS is void in USA.

or misuse of this substance.

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